

# Interface layer formed in a low-temperature plasma of hydrocarbons on iron

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## Abstract

© Published under licence by IOP Publishing Ltd. The method of uncovering the interphase layer in the systems "iron-nanosized polymer film" obtained in a low-temperature plasma on iron from hydrocarbons was used. A method of interface layer uncovering based on electrochemical dissolution of the iron substrate was proposed, which allowed one to achieve a high rate of metal. Ammonium sulfate solution was used as an electrolyte. A method of interface layer uncovering based on electrochemical dissolution of the iron substrate was proposed in the work. This allowed one to achieve a high rate of metal dissolution and make it possible to control the composition of the dissolution product. Salt solution was used as an electrolyte. Interface layer studies were carried out using the AFM and XPS methods, which allowed establishing the relief and chemical composition of the interphase boundary, as well as the location on the interphase surface of the metal-containing phase. The application of a double-layer iron substrate consisting of a plate of 56 Fe nonresonance isotope with a 57 Fe ultrafine layer deposited on its surface allowed one to use the Mössbauer spectroscopy for investigation of the chemical state of iron atoms in the interfacial region of "iron-nanoscale polymer film" systems.

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